AL-FARABI KAZAKH NATIONAL UNIVERSITY

Faculty of Chemistry and Chemical Technology
Department of Chemistry and Technology of
Organic Substances, Natural Compounds and
Polymers

	coved of the Faculty	
		Ongarbaev E.K
"	"	2017

SYLLABUS

fall semester, 2017-2018 academic year

Academic Information about the Course

Code of the	The name of	Type	No. of hours per week			Number	ECTS	
discipline	discipline		Lecture	Practic	e	Laboratory	of credits	
						work		
BH 3419	Bioorganic	ОК	1,5	0		3	4,5	7,5
	Chemistry							
Lecturer	Janar Jenis, Ph.D.			O	ffice Hours	According	to the	
e-mail	janarjenis@mail.ru					timetab	le	
Phone (mob.)	87016677659			O	ffice Room	525		
Assistant	Gadetskaya Anastassiya, Ph.D.			Office Hours		According to the		
e-mail	avg01.08@mail.ru					timetab	le	
Phone (mob.)	87017470048				O	ffice Room	515	

Academic	Type of training course:			
presentation of	Intended to give insight into the specific and fundamental role of organic			
the course	reactions occurring in nature, to provide students with a basic understanding of			
	the chemical nature of biomolecules and biomacromolecules. The emphasis will			
	be on the patterns of reactivity among natural products, rather than on the			
	biochemical roles that these molecules play.			
	Goal of the course is to gain familiarity with basic chemical principles,			
	especially as they relate to biological systems.			
Prerequisites	Organic Chemistry, Physical Chemistry, Catalysis			
Postrequisites	Chemistry of Cyclic Compounds, Chemical Technology of Organic Substances			
Informational	Study literature:			
resources	1. H. Stephen Stoker General, Organic, and Biological Chemistry 5th Edition.			
	2. Kenneth W. Raymond General, Organic, and Biological Chemistry: An			
	Integrated Approach 2nd Edition.			
	3. H. Stephen Stoker Organic, and Biological Chemistry 2nd Edition.			
	4. David Van Vranken and Gregory Weiss Introduction to Bioorganic Chemistry			
	and Chemical Biology			
	5. Donald Voet, Judith G. Voet Biochemistry, 4th Edition.			
	6. Ch. Pratt and K. Cornly, Essential Biochemistry, 3d edition.			

	T							
	7. Fromm, Herbert J., Hargrove, Mark Essentials of Biochemistry, 2012th							
	edition.							
	Internet resources:							
	1. www.chem.qmul.ac.uk/iubmb							
	2. <u>www.chemport.org</u>							
	3. <u>www.febs.org</u>							
	4. <u>www.molbiol.ru</u>							
	5. www.ncbi.nlm.nih.gov/Genbank							
	6. www.swissprot.com							
	7. www.ncbi.nlm.nih.gov/PubMed							
	8. http://www.lipidlibrary.co.uk/lipids.html							
	9. http://www.genome.jp							
Academic	Rules of academic beh	avior:						
policy of the	Plagiarism and other fo	rms of cheating, cribb	oing and hints during the surrender					
course in the	of ISW, interim monito	ring, as well as the fir	nal examination are prohibited.					
context of	The student, caught on	the falsifying of any i	nformation will receive a final					
university	grade «F».							
values	For the consultations of	n ISW, exams and oth	er questions, please contact the					
	teacher during office ho		1					
	Academic values:							
	Be tolerant and respect the other people's opinion. Objection must be formulated							
	in the correct form.		·					
Evaluation and	Criterial evaluation:							
attestation	Description of the wor	k Percentage	Completion time					
policy	Laboratory work	70%	1 - 15					
	Assessment	30%	7, 14					
	Exam	40%	8,16					
	TOTAL	100%	,					
	Summative estimation	ı :						
The final score will be calculated by the formula: Final grade = 0,3 (Short Exam 1 + Short Exam 2) + 0,1 Midterm Exam + Final Exam								
				Evaluation scheme in percentage:				
					95% - 100%: A	_	85% - 89%: B+	
	80% - 84%: B	75% - 79%: B-	70% - 74%: C+					
	65% - 69%: C	60% - 64%: C-	55% - 59%: D+					
	50% - 54%: D-	0% -49%: F						

$Calendar\ for\ the\ implementation\ of\ the\ training\ course\ content$

Week/	Topic title (lecture, practical classes, ISW)	No. of	Maximum
date		hours	Score
1	2	3	5
1	Lecture 1. Introduction.	1	7
	Laboratory 1. Laboratory Safety.	2	
2	Lecture 2. Overview of bioorganic chemistry.	1	13
	Laboratory 2. Preparation of acetyl salicylic acid.	2	
	ISWT: Submission of the task 1		
	(Nomenclature of main biomolecules)		

3	Lecture 3. Lipids.	1	8
	Laboratory 3. Preparation of acetyl salicylic acid. Continued.		
4	Lecture 4. Lipids. Continued.	1	13
	Laboratory 4. Extraction of a known mixture.	2	
	ISWT: Submission of the task 2		
	(Difference in use of detergents based on)		
5	Lecture 5. Carbohydrate.	1	8
	Laboratory 5. Extraction of a known mixture. Continued.	2	
6	Lecture 6. Carbohydrate. Continued.	1	13
	Laboratory 6. Extraction of an unknown mixture.	2	
	ISWT: Submission of the task 3		
	(Functional groups in biological molecules)		
7	Lecture 7. Amino acids.	1	8
	Laboratory 7. Extraction of an unknown mixture. Continued.	2	
	Assessment (Short Exam 1)		30
8	Lecture 8. Peptides.	1	7
	Laboratory 8. Synthesis of acetanilide.	2	
9	Lecture 9. Proteins. Primary structure.	1	10
	Laboratory 9. Synthesis of acetanilide. Continued.	2	
	ISWT: Submission of the task 4		
	(Analysis of lidocaine)		
10	Lecture 10. Three-Dimensional Structure of Proteins. Protein	1	8
	Folding.		
	Laboratory 10. Recrystallization of acetanilide.	2	
11	Lecture 11. Nucleic acids.	1	10
	Laboratory 11. Recrystallization of acetanilide. Continued.	2	
	ISWT: Submission of the task 5		
	(Molecular diseases)		
12	Lecture 12. Nucleic acids. Continued.	1	7
	Laboratory 12. The synthesis of soap.	2	
13	Lecture 13. Genomics and Proteomics	1	10
	Laboratory 13. The synthesis of soap. Continued.	2	
	ISWT: Submission of the task 6		
	(Cell structure and organization)		
14	Lecture 14. Enzymes.	1	8
	Laboratory 14. Essential oils of plants.	2	
	Assessment (Short Exam 2)		30
15	Lecture 15. Enzyme Mechanisms.	1	10
	Laboratory 15. Essential oils of plants. Continued.	2	
	ISWT: Submission of the task 7		
	(Protein Sequencing)		

Lecturer	Jenis J.		
Head of the Department		_ Mun G.A.	
Chairman of the Methodical Bureau of t	he Faculty		Ongarbaev E.K.